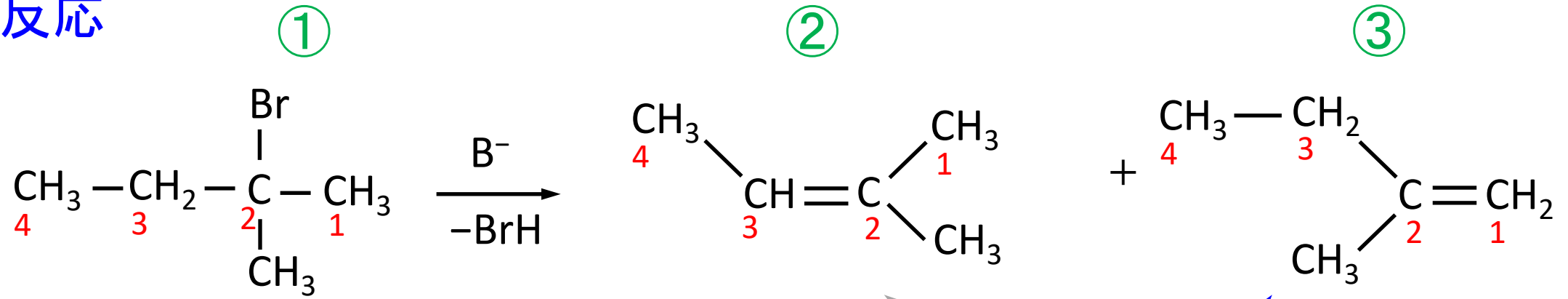
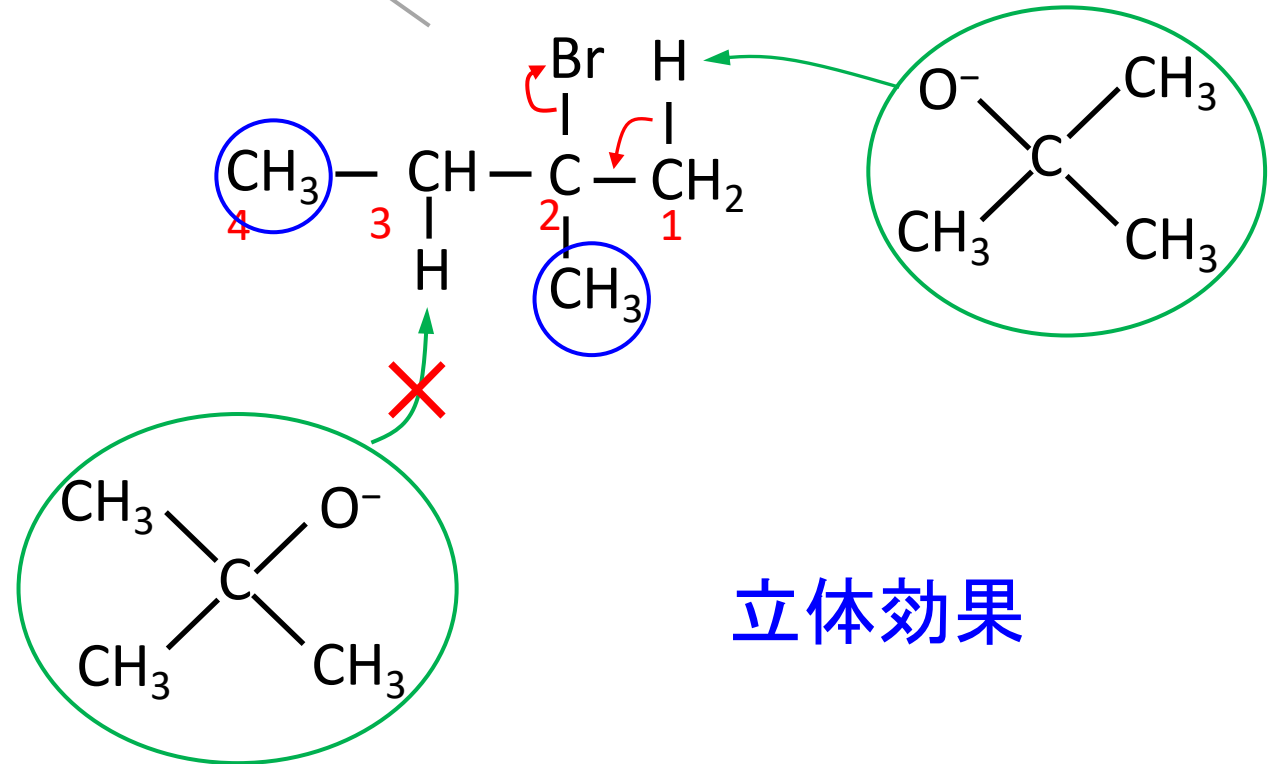
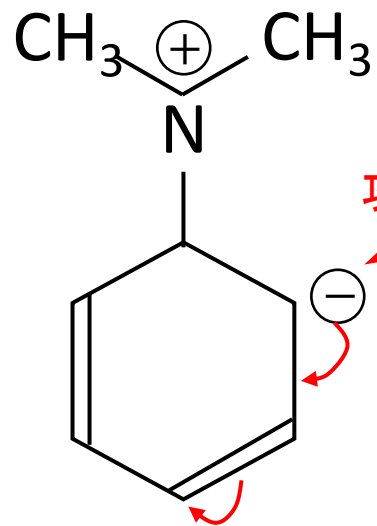
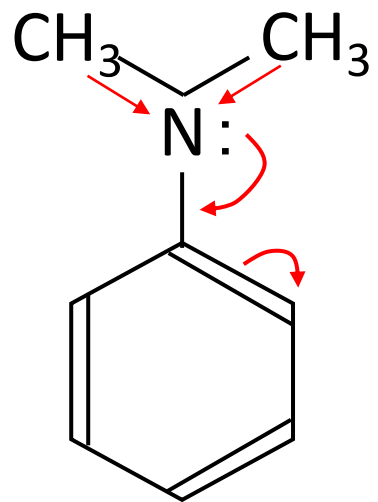


# 脱離反応



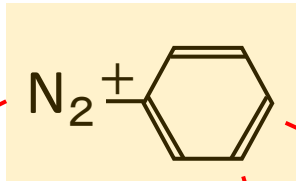
$\text{B}^-$	② : ③	
$\text{CH}_3\text{---CH}_2\text{---O}^-$	70 : 30	ザイツェフ則
$\text{CH}_3\text{---C(CH}_3\text{)}_2\text{---O}^-$	27 : 73	ホフマン則





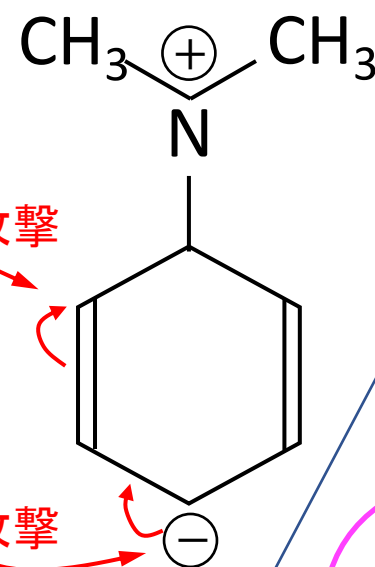
ジアゾカップリング

攻撃

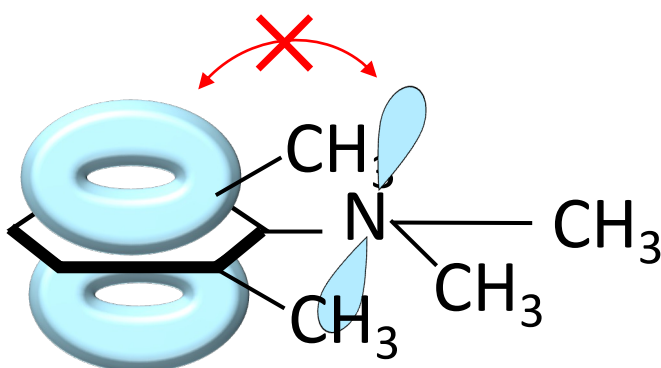
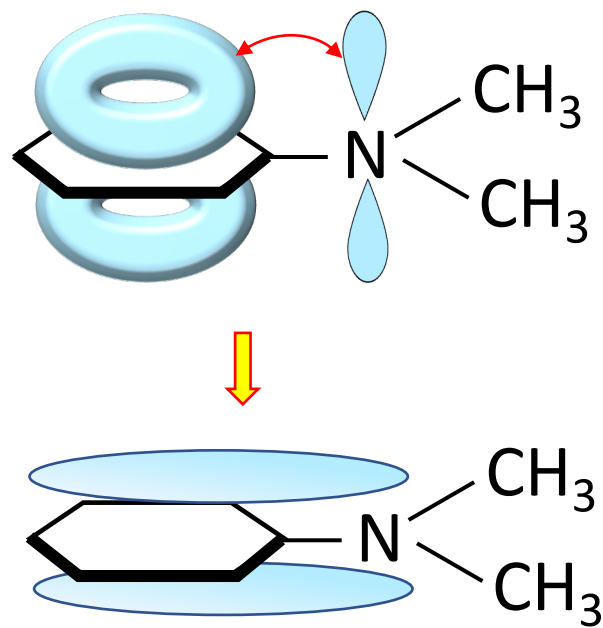


攻撃

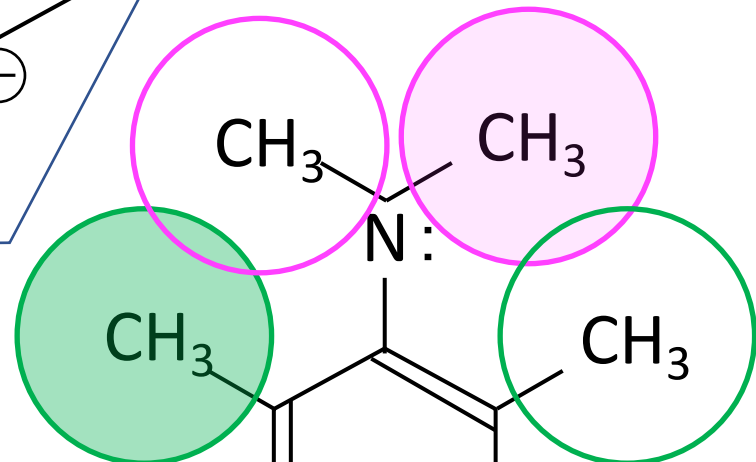
攻撃



立体効果



4つのCH<sub>3</sub>基が同一平面になれないため、窒素のp軌道がベンゼン環のπ電子と非局在化できない



攻撃できない

